

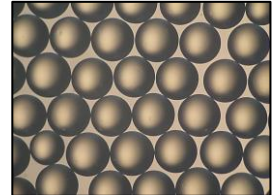


AMBERLITE™ HPR550 Cl Ion Exchange Resin

Uniform Particle Size, Gel, Strong Base Anion Exchange Resin for Condensate Polishing and Mixed Bed Demineralization Applications for the Power Industry

Description

AMBERLITE™ HPR550 Cl Ion Exchange Resin is a premium-quality, high-capacity resin specifically designed for use in high purity industrial demineralization applications and condensate polishing beds at power stations when the chloride-form is preferred by the user.



This resin provides exceptional bead integrity and rapid exchange kinetics due to its small average particle size, making it ideally suited to the high flowrate demands commonly encountered in power plant condensate polishing systems. The bead size uniformity and a distinguishable light color is tailored to complement the larger, denser, cationic, gel AMBERLITE™ HPR650 H Ion Exchange Resin. The color distinction between this pair of resins allows easy visual confirmation of separation following backwash.

For post-RO mixed bed polishing with a strict silica specification and/or the need to maximize silica removal capacity, AMBERLITE HPR550 Cl is an alternative to the OH-form.

Resin Pairings

Recommended pairing:

- AMBERLITE™ HPR650 H Ion Exchange Resin (gel)

Additional options:

- AMBERLITE™ HPR1600 H Ion Exchange Resin (gel)
- AMBERLITE™ HPR2000 H Ion Exchange Resin (macroporous) – in external regeneration systems

Applications

- Mixed bed condensate polishing in fossil power plants
- Mixed bed polishing in industrial demineralization
- Demineralization
 - Ideally when treating water with:
 - High percentage of silica
 - When the treatment goal is:
 - Removal of strong and weak acids
 - Lowest silica leakage
 - Single bed industrial demineralization requiring high water purity

Historical Reference

AMBERLITE™ HPR550 Cl Ion Exchange Resin has previously been sold as DOWEX MONOSPHERE™ 550A Ion Exchange Resin.

Typical Physical and Chemical Properties**

Physical Properties	
Copolymer	Styrene-divinylbenzene
Matrix	Gel
Type	Strong base anion
Functional Group	Trimethylammonium
Physical Form	White to yellow, translucent, spherical beads

Chemical Properties	
Ionic Form as Shipped	Cl ⁻
Total Exchange Capacity	≥ 1.35 eq/L (Cl ⁻ form)
Water Retention Capacity	42.0 – 49.0% (Cl ⁻ form)

Particle Size	
Particle Diameter §	550 ± 50 μm
Uniformity Coefficient	≤ 1.1
< 300 μm	≤ 0.5%
> 850 μm	≤ 1.0%

Stability	
Whole Uncracked Beads	≥ 95%
Friability:	
Average	≥ 350 g/bead
> 200 g/bead	≥ 95%
Swelling	Cl ⁻ → OH ⁻ ≤ 25%

Density	
Particle Density	1.09 g/mL
Shipping Weight	690 g/L

§ For additional particle size information, please refer to the [Particle Size Distribution Cross Reference Chart](#) (Form No. 177-01775).

Suggested Operating Conditions**

Temperature Range (OH ⁻ form) †	5 – 100°C (41 – 212°F)
pH Range (Stable)	0 – 14

† Operating at elevated temperatures, for example above 60 – 70°C (140 – 158°F), may impact the purity of the loop and resin life. Contact our technical representative for details.

For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for [mixed beds](#) (Form No. 177-03705) or [separate beds](#) (Form No. 177-03729) in water treatment, please refer to our Tech Facts.

Hydraulic Characteristics

Estimated bed expansion of AMBERLITE™ HPR550 Cl Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Estimated pressure drop for AMBERLITE HPR550 Cl as a function of service flowrate and temperature is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean water.

Figure 1: Backwash Expansion

Temperature = 10 – 60°C (50 – 140°F)

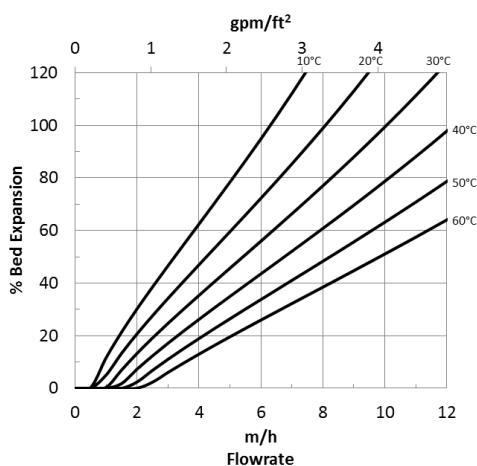
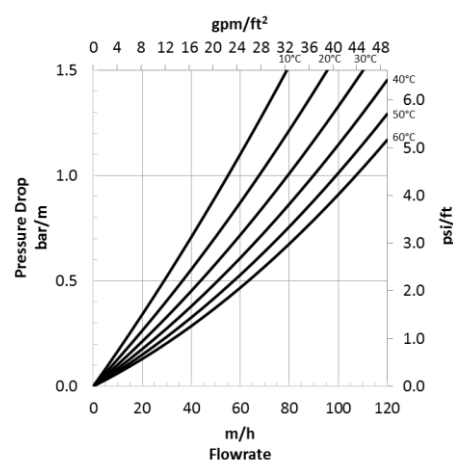


Figure 2: Pressure Drop

Temperature = 10 – 60°C (50 – 140°F)



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www.dowwaterandprocess.com

WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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