

Product Data Sheet

## AMBERLITE<sup>™</sup> HPR900 OH Ion Exchange Resin

Macroporous, Strong Base Anion Exchange Resin for Condensate Polishing for the Power Industry and Industrial Demineralization Applications

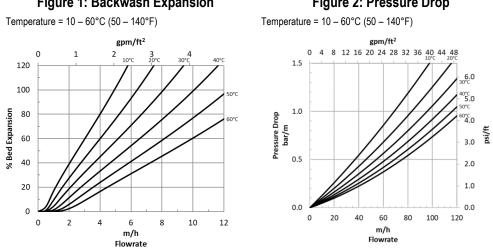
Description	AMBERLITE <sup>™</sup> HPR900 OH Ion Exchange Resin is specifically designed for use in condensate polishing beds at fossil-fired electric generating stations and industrial demineralization applications when a balance of operating performance, simple operation, long resin life, and cost-effective operation is required.	
	The macroporous structure of AMBERLITE HPR900 OH provides resistance to surface fouling as well as physical, osmotic, and oxidative stresses, which allows increased resin lifetime in operation. The resin can operate reliably under the high flowrate and pressure drop conditions that are typically inherent in condensate polishers.	
	This resin is designed to be used in combination with AMBERLITE™ HPR252 H lon Exchange Resin and AMBERLITE™ 600i Inert Resin in TRIOBED™ Condensate Polishers, providing an optimized balance of stability, operating capacity, low pressure drop, and regeneration efficiency.	
	When high water quality and long runtime are needed, AMBERLITE™ HPR1300 H lon Exchange Resin is a trusted choice.	
Resin Pairings	<ul> <li>Recommended pairing:</li> <li>AMBERLITE™ HPR252 H Ion Exchange Resin (macroporous)</li> <li>AMBERLITE™ HPR1300 H Ion Exchange Resin (gel)</li> <li>AMBERLITE™ HPR2800 H Ion Exchange Resin (macroporous)</li> </ul>	
Applications	<ul> <li>Mixed bed condensate polishing in fossil power plants</li> <li>Mixed bed polishing in industrial demineralization</li> <li>Systems requiring exceptionally high osmotic stability</li> </ul>	
Historical Reference	AMBERLITE™ HPR900 OH Ion Exchange Resin has previously been sold as AMBERSEP™ 900 OH Ion Exchange Resin.	

## Typical Physical and Chemical Properties<sup>\*\*</sup>

Physical Properties	
Copolymer	Styrene-divinylbenzene
Matrix	Macroporous
Туре	Strong base anion
Functional Group	Trimethylammonium
Physical Form	White, opaque, spherical beads
Chemical Properties	
Ionic Form as Shipped	OH-
Total Exchange Capacity	≥ 0.80 eq/L (OH <sup>−</sup> form)
Water Retention Capacity	66.0 – 75.0% (OH <sup>-</sup> form)
Ionic Conversion	
OH-	≥ 95%
CO3 <sup>2-</sup>	≤ 5%
CI⁻	≤ 0.50%
Particle Size	
Particle Diameter §	500 – 700 μm
Uniformity Coefficient	≤ 1.45
< 300 µm	≤ 0.5%
> 1180 µm	≤ 5.0%
Stability	
Whole Uncracked Beads	≥ 96%
Swelling	$CI^- \rightarrow OH^- \le 25\%$
Density	
Particle Density	1.05 g/mL
Shipping Weight	675 g/L

§ For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 177-01775).

Suggested Operating Conditions**	Temperature Range (OH <sup>-</sup> form) <sup>‡</sup> pH Range (Stable)	5 – 100°C (41 – 212°F) 0 – 14	
	<sup>‡</sup> Operating at elevated temperatures, for exampl life. Contact our technical representative for del	e above 60 – 70°C (140 – 158°F), may impact the purity of the loop and resin tails.	
	For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for <u>mixed beds</u> (Form No. 177-03705) or <u>separate</u> <u>beds</u> (Form No. 177-03729) in water treatment, please refer to our Tech Facts.		
Hydraulic Characteristics	Estimated bed expansion of AMBERLITE™ HPR900 OH Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.		
	Estimated pressure drop for AMBERLITE HPR900 OH as a function of service flowrate and temperature is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean water and a well-classified bed.		
	Figure 1: Backwash Expans	ion Figure 2: Pressure Drop	



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## For more information, contact our Customer Information Group:

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**WARNING:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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